Teacher: L. George	Week of: January 2-6, 2017
Subject: 8 <sup>th</sup> Grade General Science	Week 1 Quarter 3

Monday	Tuesday	Wednesday	Thursday	Friday
SPI/Check for	SPI/Check for	SPI/Check for	SPI/Check for	SPI/Check for
Understanding:	Understanding:	Understanding:	Understanding:	Understanding:
	<ul> <li>GLE 0807.9.1 Understand that all matter is made up of atoms.</li> <li>GLE 0807.9.7 Explain the Law of Conservation of Mass GLE 0807.9.8 Interpret the events represented by a chemical equation.</li> </ul>	<ul> <li>GLE 0807.9.1 Understand that all matter is made up of atoms.</li> <li>GLE 0807.9.7 Explain the Law of Conservation of Mass GLE 0807.9.8 Interpret the events represented by a chemical equation.</li> </ul>	<ul> <li>GLE 0807.9.1 Understand that all matter is made up of atoms.</li> <li>GLE 0807.9.7 Explain the Law of Conservation of Mass</li> <li>GLE 0807.9.8 Interpret the events represented by a chemical equation.</li> </ul>	<ul> <li>GLE 0807.9.1 Understand that all matter is made up of atoms.</li> <li>GLE 0807.9.7 Explain the Law of Conservation of Mass</li> <li>GLE 0807.9.8 Interpret the events represented by a chemical equation.</li> </ul>
Students will understand.	Students will understand.	Students will understand.	Students will understand.	Students will understand.
know, be able to	<ul> <li>know, be able to</li> <li>Interpret and write simple chemical formulas</li> <li>Write and balance simple chemical</li> </ul>	<ul> <li>know, be able to</li> <li>Describe 4 types of reactions</li> <li>Classify a chemical equation as one of four types of chemical</li> </ul>	<ul> <li>know, be able to</li> <li>Interpret and write simple chemical formulas</li> <li>Write and balance simple chemical</li> </ul>	<ul> <li>know, be able to</li> <li>Describe 4 types of reactions</li> <li>Classify a chemical equation as one of four types of chemical</li> </ul>
	equations Explain how a balanced equation shows the law of conservation of mass	Compare exothermic reactions and endothermic reactions	<ul> <li>equations</li> <li>Explain how a balanced equation shows the law of conservation of mass</li> </ul>	<ul> <li>Compare exothermic reactions and endothermic reactions</li> </ul>
Learning Activities	<ul> <li>Learning Activities</li> <li>Review Notes on Balancing and Parts of Equations</li> <li>Chapter 13 Vocabulary</li> </ul>	<ul> <li>Learning Activities</li> <li>Review chapter 13.2</li> <li>Worksheet: balancing act</li> <li>HW: Finish worksheets from class</li> <li>Introduce Project on Types of Reactions (Lab)</li> </ul>	<ul> <li>Learning Activities</li> <li>Notes: Types of Chemical reactions Review</li> <li>Worksheet: Chemical reactions</li> <li>Project: Booklet Types of Reactions</li> <li>HW: page 347 #2-6</li> </ul>	<ul> <li>Learning Activities</li> <li>Notes: Types of Chemical Reaction</li> <li>Chapter 13 Vocabulary Quiz</li> </ul>
Assignments	Assignments <ul> <li>Review notes</li> <li>HW: Complete and <ul> <li>Study Chapter 13</li> <li>Vocabulary</li> </ul> </li> </ul>	Assignments HW: Complete Balancing Act Types of Reaction Project due Jan. 11th	Assignments • HW: Study for Chapter 13 Vocabulary Quiz	Assignments • HW: Review for Quiz Balancing Equations and Types of Reactions on Tuesday